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## Diastereoselective arylation of L-proline derivatives at the 5-position

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**Abstract-** Diastereoselective introduction of nucleophiles into L-proline derivatives at the 5-position was achieved with suitable selection of N-protecting group. N-Methoxycarbonylated or benzyloxycarbonylated L-proline derivatives reacted with arene to give cis-arylated products. On the other hand, N-benzoylated L-proline derivative preferentially gave trans-arylated product which could be easily transformed into optically active  $C_2$ -symmetrical pyrrolidine derivative. Such derivative  $\mathbf{5}$ , worked well as an organic activator in the asymmetric reduction of aromatic imines by  $Cl_3SiH$ .

*Keywords:* Diastereoselective; Organocatalysis; Asymmetric reduction of imines; C<sub>2</sub>-Symmetrical pyrrolidine; Proline derivative

#### 1. Introduction

Optically active 2,5-disubstituted pyrrolidines are key intermediates for preparation of pharmaceuticals or natural products<sup>1</sup> as well as organocatalysts for asymmetric reactions.<sup>2</sup> Electrochemical oxidation of L-proline derivatives **1** is a useful tool for their synthesis (Eq. 1).<sup>3</sup>

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Recently, we have reported that cis-5-arylated N-formyl-L-proline  $\mathbf{4}^4$  worked well as an organic activator in the enantioselective reduction of ketones with  $Cl_3SiH^5$  in high enantioselectivities (Eq. 2). However, it was difficult to prepare  $\mathbf{4}$  for practical use because diastereoselectivity in arylation reaction of  $\mathbf{2a}$  (PG=CHO) was very low. We wish herein to report diastereoselective introduction of nucleophiles into L-proline derivatives at the 5-position. In addition, synthesis of compound  $\mathbf{4}$  and  $C_2$ -symmetrical pyrrolidine derivative  $\mathbf{5}$  derived from cis- and trans-arylated products, and its application to asymmetric reduction of aromatic imines with  $Cl_3SiH^6$  are presented.

### 2. Results and discussion

2.1. Diastereoselective introduction of nucleophiles into L-proline derivatives at the 5-position

First, we investigated introduction of trimethylbenzene and triethylbenzene into 5-methoxylated L-proline derivatives **2a-d**<sup>7</sup> protected with various *N*-acyl groups in the presence of Lewis acids (Eq. 3). The results are shown in Table 1. *N*-Formylated proline **2a** gave the corresponding arylated product **3a** as a diastereomer mixture (*cis/trans* = 43/57, entry 1),<sup>4</sup> while *N*-methoxycarbonylated **2b** and *N*-benzyloxycarbonylated **2c** gave compounds **3b** and **3c** as a single isomer (*cis/trans* = 100/0, entries 2 and 3). In the case of *N*-benzoylated **2d**, *trans*-**3d**<sup>8</sup> was mainly obtained along with small amount of *cis*-**3d** (*cis/trans* = 11/89, entry 4). Using SnCl<sub>4</sub> instead of TiCl<sub>4</sub> did not affect the diastereoselectivity though the former had relatively poor yield (entry 5). Triethylbenzene as a nucleophile gave similar results to that of trimethylbenzene (entries 6-9), but in the case of *N*-benzoylated proline **2d** did not afford 5-arylated product **6d** (entry 10).

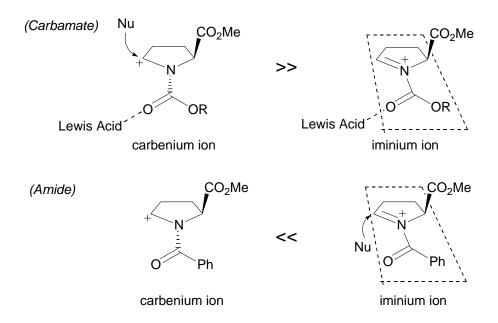
Table 1. Arylation of proline derivative 2a-d at the 5-position

Entry	PG		Lewis acid	R	Yield (%)	cis/trans	
1	CHO	2a	TiCl <sub>4</sub>	Ме	<b>3a</b> 61	43/57	
2	CO <sub>2</sub> Me	2b	TiCl <sub>4</sub>	Me	<b>3b</b> 51	100/0	
3	Cbz	2c	TiCl <sub>4</sub>	Me	<b>3c</b> 68	100/0	
4	Bz	2d	TiCl <sub>4</sub>	Me	<b>3d</b> 65	11/89	
5	Bz	2d	SnCl <sub>4</sub>	Me	<b>3d</b> 43	11/89	
6	CHO	2a	SnCl <sub>4</sub>	Et	<b>6a</b> 71	52/48	
7	CO <sub>2</sub> Me	2b	SnCl <sub>4</sub>	Et	<b>6b</b> 55	100/0	
8	Cbz	2c	SnCl <sub>4</sub>	Et	<b>6c</b> 36	100/0	
9	Cbz	2c	TiCl <sub>4</sub>	Et	<b>6c</b> 31	100/0	
10	Bz	2d	SnCl <sub>4</sub>	Et	<b>6d</b> 0	-	

Allylation of 5-methoxylated L-proline derivatives **2b** and **2d** showed similar tendency to their arylation (Eq. 4). That is, *N*-methoxycarbonylated **2b** mainly gave *cis*-allylated proline **7b** (cis/trans = 73/27), <sup>7c</sup> while *N*-benzoylated proline **2d** preferentially changed into *trans*-allylated proline **7d** (cis/trans = 13/87). <sup>9</sup>

MeO 
$$\stackrel{\text{N}}{\text{PG}}$$
 CO<sub>2</sub>Me  $\stackrel{\text{allyl-SiMe}_3}{\text{CO}_2\text{Me}}$  (3.0 equiv for **2b**, 1.0 equiv for **2d**)  $\stackrel{\text{N}}{\text{CO}_2\text{Me}}$  (4)  $\stackrel{\text{TiCl}_4}{\text{Cl}_2\text{Cl}_2}$ , -78°C to 0°C, 12h  $\stackrel{\text{PG}}{\text{CO}_2\text{Me}}$  PG = CO<sub>2</sub>Me **7b** : 80% (cis/trans = 73/27) = Bz  $\stackrel{\text{7d}}{\text{7d}}$  : 53% (cis/trans = 13/87)

Key intermediates in these reactions are carbenium and iminium ions illustrated in Scheme 1. Since the carbonyl group of carbamates (PG= CO<sub>2</sub>Me or Cbz) can coordinate to Lewis acid, carbenium ion will be preferable to iminium ion. On the other hand, carbonyl group of amide (PG=Bz) might not coordinate to Lewis acid. Therefore, the iminium ion will be predominantly generated. The *cis*-selectivity in the carbenium ion intermediate is illustrated in Scheme 1 (*Carbamate*) in which PG (CO<sub>2</sub>Me or Cbz) is oriented in *trans* position with respect to 2-CO<sub>2</sub>Me substituent. Nucleophiles may approach the intermediate preferentially from the *trans* direction with respect to PG.<sup>7c</sup> The *trans*-selectivity (PG=Bz) in the iminium ion intermediate is illustrated in Scheme 1 (*Amide*) in which Bz and iminium groups exist on the same plane. Nucleophiles can approach the intermediate preferentially from the *trans* direction with respect to 2-CO<sub>2</sub>Me substituent.



Scheme 1. Plausible stereochemical course.

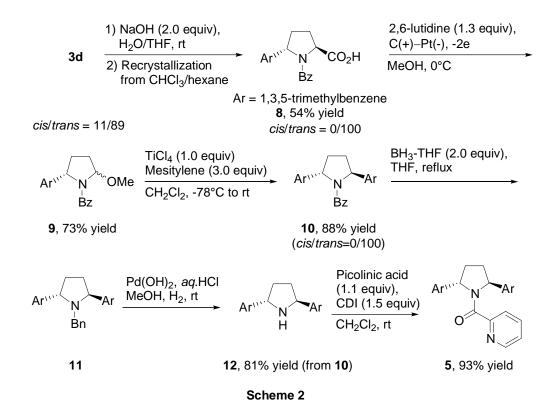
## 2.2. Synthesis of an organic activator 4 and C2-symmetrical pyrrolidine derivative 5

An organic activator **4** for the enantioselective reduction of ketones was synthesized from **6c** after hydrogenation, *N*-formylation followed by alkaline hydrolysis in 58% yield (Eq. 5).<sup>10</sup>

Et 
$$CD_2Me$$
 1) Pd-C,  $H_2$ ,  $Et_3N$ ,  $MeOH$  2)  $Ac_2O$ ,  $HCO_2H$  3) NaOH,  $H_2O/THF$  Et  $CHO$  Et  $CHO$  4 58% yield

 $C_2$ -Symmetrical pyrrolidine derivative **5** was prepared from *N*-benzoylated proline **3d** as follows (Scheme 2); Alkaline hydrolysis of **3d** followed by recrystallization from CHCl<sub>3</sub>/hexane afforded carboxylic acid **8** in 54% yield as a single isomer (cis/trans = 0/100). Electrochemical decarboxylative methoxylation<sup>11</sup> of **8** in methanol afforded methoxylated compound **9**, which reacted with mesitylene in the presence of TiCl<sub>4</sub> to

exclusively afford *trans*-2,5-biarylated pyrrolidine **10** in high yield. By reduction of *N*-benzoyl group of **10**, successive deprotection of *N*-benzyl group of **11**, and *N*-picolynoylation of **12**, desired pyrrolidine **5** was obtained in enough yield.



### 2.3. Asymmetric reduction of aromatic imines catalyzed by 5 with Cl<sub>3</sub>SiH

Catalytic activation of Cl<sub>3</sub>SiH with compound **5** was applicable to asymmetric reduction of aromatic imines **13a-f** (Eq. 6). The results are summarized in Table 2, which also shows the results of asymmetric reduction using **15**<sup>6c</sup> for comparison. In all cases, compound **5** could play the role of an activator to afford (*S*)-amines **14a-f**<sup>6</sup> with good yield and enantioselectivity just like that of **15** (entries 1-6).

Table 2. Asymmetric reduction of imines 13a-f

Entry	Imine	R <sup>1</sup>	$R^2$	(S)-Amine	Activator 5		Activator 15	
					Yield (%)	ee (%) <sup>a</sup>	yield (%)	ee (%) <sup>a</sup>
1	13a	Н	Н	14a	92	77	86	73
2	13b	Н	OMe	14b	84	78	90	71
3	13c	OMe	Н	14c	87	76	90	75
4	13d	Н	CI	14d	88	73	73	71
5	13e	Н	Ac	14e	60	64	24	67
6	13f	Н	$NO_2$	14f	74	85	84	73

<sup>&</sup>lt;sup>a</sup> Determined by HPLC.

### 3. Conclusion

We have accomplished diastereoselective introduction of nucleophiles into L-proline derivatives at the 5-position. N-Methoxycarbonylated or N-benzyloxycarbonylated L-proline **2b** or **2c** were exclusively transformed into cis-arylated products **3b**,c or **6b**,c, while N-benzoylated L-proline derivative **2d** mainly gave trans-arylated product **3d**.  $C_2$ -Symmetrical pyrrolidine derivative **5** derived from **3d** worked well as an organic activator in the reduction of aromatic imines to the corresponding optically active amines with high enantioselectivity by  $Cl_3SiH$ .

## 4. Experimental Section

### 4.1. General

Electrochemical reactions were carried out using DC Power Supply (GP 050-2) of Takasago Seisakusho, Inc. <sup>1</sup>H NMR spectra were measured on a Varian Gemini 300 and 400 spectrometer with TMS as an internal standard. <sup>13</sup>C NMR spectra were measured on a Varian Gemini 400 spectrometer with TMS as an internal standard. IR spectra were obtained on a Shimadzu FTIR-8100A. Mass spectra were obtained on a JEOL JMS-DX 303 instrument. Elemental analyses were performed on Perkin Elmer 2400II.

All reagents and solvents were used as supplied without further purification.

### 4.2. Methyl N-protected 5-methoxy-L-prolinate 2a-d

*N*-Protected 5-methoxy-L-prolinates  $2\mathbf{a}$ ,  $^{7c}$   $2\mathbf{b}$ ,  $^{7a}$   $2\mathbf{c}$ ,  $^{7d}$  and  $2\mathbf{d}^{7b}$  were known compounds.

4.3. General procedure for arylation or allylation of methyl N-protected-5-methoxy-L-prolinate **2a-d** 

Under an argon atmosphere, TiCl<sub>4</sub> (55  $\mu$ L, 0.5 mmol) was added dropwise to the solution of **2a** (109 mg, 0.5 mmol) and 1,3,5-trimethylbenzene (209  $\mu$ L, 1.5 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (5 mL) at -78°C. The resulting mixture was stirred for 12 h and allowed to stand until it warmed to room temperature. The solution was poured in ice water (10 mL) and extracted with CHCl<sub>3</sub> (10 mL x 3). The combined organic layer was dried over MgSO<sub>4</sub> and the solvent removed under reduced pressure. The residue was purified by silica gel column chromatography (*n*-hexane : AcOEt = 10 : 1) to afford **3a** as a colorless oil (93 mg, 61 %). Arylation with 1,3,5-triethylbenzene and allylation with allyltrimethylsilane were carried out according to this same procedure.

4.3.1. Methyl cis-N-formyl-5-(2,4,6-trimethylphenyl)-L-prolinate (cis-3a)<sup>4</sup>

Colorless oil; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.85 (s, 1H), 6.88 (s, 2H), 5.04 and 5.06 (d, J= 11.0 Hz, 1H), 4.53 (t, J= 5.4 Hz, 1H), 3.80 (s, 3H), 2.55-1.95 (m, 13H).

4.3.2. Methyl trans-N-formyl-5-(2,4,6-trimethylphenyl)-L-prolinate (trans-3a)<sup>4</sup>

Colorless oil; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.81 (s, 1H), 6.85 (s, 2H), 5.37 (t, J= 8.0 Hz, 1H), 4.56 (t, J= 7.5 Hz, 1H), 3.75 (s, 3H), 2.42-1.99 (m, 13H).

## 4.3.3. Methyl N-methoxycarbonyl-5-(2,4,6-trimethylphenyl)-L-prolinate (3b)

Colorless crystal; mp 48-50°C;  $[\alpha]^{27}_D$  -49.1 (c=1.0, CHCl<sub>3</sub>); IR (neat) v = 2953, 1754, 1701, 1612, 1447, 1348, 1198, 1123, 1078, 851, 781 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  6.80 (s, 2H), 5.10 (t, J= 9.0 Hz, 1H), 4.59-4.51 (m, 1H), 3.78 (s, 3H), 3.55 (s, 3H), 2.44-2.07 (m, 13H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  172.5, 157.1, 135.8, 135.6, 133.1, 130.0, 60.4, 52.6, 51.9, 30.2, 27.9, 20.5; HR-EI(+) m/z calcd for  $C_{17}H_{23}NO_4[M]^+$  305.1627, found 305.1623.

### 4.3.4. Methyl N-benzyloxycarbonyl-5-(2,4,6-trimethylphenyl)-L-prolinate (3c)

Colorless oil;  $[\alpha]_D^{27}$  -49.6 (c=1.0, CHCl<sub>3</sub>); IR (neat)  $\nu$  = 2960, 1753, 1701, 1456, 1338, 1197, 1174, 1120, 851, 735 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.35-7.10 (m, 5H), 6.79 (s, 2H), 5.15-4.90 (m, 3H), 4.63-4.55 (m, 1H), 3.74 (s, 3H), 2.45-2.05 (m, 13H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  172.5, 156.4, 135.8, 135.7, 133.2, 131.3, 130.1, 129.3, 128.2, 127.9, 127.8, 127.4, 127.2, 67.0, 60.4, 52.0, 30.4, 27.8, 20.6, 20.5; HR-EI(+) m/z calcd for  $C_{23}H_{27}NO_4[M]^+$  381.1940, found 381.1938.

### 4.3.5. *Methyl trans-N-benzoyl-5-(2,4,6-trimethylphenyl)-L-prolinate (3d)*

Colorless crystal; mp 112-114 °C;  $[\alpha]_D^{18}$ -133.5 (c=1.0, CHCl<sub>3</sub>); IR (neat) v = 2953, 1755, 1745, 1659, 1641, 1632, 1580, 1444, 1414, 1279, 1202, 1175, 1127, 1028, 853 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) (cis/trans = 11/89)  $\delta$  7.50-7.00 (m, 5H), 6.80 (br s, 0.11H), 6.64 (s, 0.89H), 6.60 (br s, 0.11H), 6.39 (s, 0.89H), 5.64 (t, J= 8.7 Hz, 0.11H), 5.40 (t, J= 8.7 Hz, 0.89H), 4.77 (t, J= 9.0 Hz, 1H), 3.83-3.76 (m, 3H), 2.58-1.95 (m, 13H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  170.8, 169.5, 134.3, 133.3, 132.1, 129.6, 127.2, 127.0, 125.4, 123.5, 59.3, 58.0, 50.3, 30.4, 26.8, 18.4, 18.3; HR-EI(+) m/z calcd for  $C_{22}H_{25}NO_3$  [M]<sup>+</sup> 351.1834, found 351.1832. HPLC: Daicel Chiralcel OJ-H column, n-hexane: isopropanol = 20: 1, wavelength: 254 nm, flow rate: 1.0 ml/min, retention time: 19.1 min (cis-3d), 23.3 min (trans-3d).

## 4.3.6. Methyl cis-N-formyl-5-(2,4,6-triethylphenyl)-L-prolinate (cis-6a)<sup>4</sup>

Colorless oil;  ${}^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.87 (s, 1H), 7.10-6.82 (br s, 2H), 5.03 and 5.01 (d, J= 12.0 Hz, 1H), 4.58-4.50 (m, 1H), 3.80 (s, 3H), 2.95-2.05 (m, 10H), 1.30-1.10 (m, 9H).

# $4.3.7.\ Methyl\ trans-N-formyl-5-(2,4,6-triethylphenyl)-L-prolinate\ (trans-\textbf{6a})^4$

Colorless oil; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.79 (s, 1H), 6.99 (s, 1H), 6.88 (s, 1H), 5.34 and 5.33 (d, J= 9.9 Hz, 1H), 4.62 (t, J= 7.8 Hz, 1H), 3.80 (s, 3H), 2.74 (q, J= 7.8 Hz, 2H), 2.65-2.40 (m, 5H), 2.38-2.01 (m, 3H), 1.30-1.10 (m, 9H).

### 4.3.8. Methyl trans-N-methoxycarbonyl-5-(2,4,6-triethylphenyl)-L-prolinate (**6b**)

Colorless oil;  $[\alpha]_D^{28}$  -41.3 (c=1.1, CHCl<sub>3</sub>); IR (neat)  $\nu$  = 2963, 1755, 1709, 1445, 1348, 1198, 1150, 1125, 1080, 874, 781 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  6.86 (m,

2H), 5.10 (t, J= 8.7 Hz, 1H), 4.60-4.50 (m, 1H), 3.80-3.50 (m, 6H), 2.80-2.11 (m, 10H), 1.30-1.10 (m, 9H);  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>) (a mixture of rotamers)  $\delta$  172.8, 157.1, 142.5, 142.1, 141.9, 141.7, 132.0, 127.0, 125.5, 60.5, 59.8, 52.1, 51.8, 32.6, 28.2, 28.0, 26.5, 25.6, 24.9, 24.7, 15.9, 15.5, 15.4, 15.2, 14.8; HR-EI(+) m/z calcd for C<sub>20</sub>H<sub>29</sub>NO<sub>4</sub> [M]<sup>+</sup> 347.2097, found 347.2081.

## 4.3.9. Methyl trans-N-benzyloxycarbonyl-5-(2,4,6-triethylphenyl)-L-prolinate (6c)

Colorless oil;  $[\alpha]_D^{28}$  -42.3 (c=1.0, CHCl<sub>3</sub>); IR (neat)  $\nu$  = 2965, 1755, 1705, 1408, 1339, 1198, 1175, 1080, 696 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.40-6.80 (m, 7H), 5.20-4.80 (m, 3H), 4.60-4.50 (m, 1H), 3.77 (s, 3H), 3.10-1.95 (m, 10H), 1.32-0.86 (m, 9H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) (a mixture of rotamers)  $\delta$  172.6, 156.7, 142.4, 142.3, 142.1, 135.7, 132.3, 128.1, 127.9, 127.7, 127.4, 126.5, 67.0, 61.0, 60.8, 59.9, 57.7, 52.2, 52.1, 33.3, 32.7, 28.9, 28.2, 27.8, 27.5, 27.1, 26.3, 25.1, 25.0, 24.8, 15.9, 15.7, 15.4, 15.3, 15.2; HR-EI(+) m/z calcd for C<sub>26</sub>H<sub>33</sub>NO<sub>4</sub> [M] + 423.2410, found 423.2394.

## 4.3.10. Methyl N-methoxycarbonyl-5-allyl-L-prolinate (7b)<sup>7c</sup>

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) (*cis/trans* = 73/27) δ 5.83-5.65 (m, 1H), 5.12-5.03 (m, 2H), 4.40-4.27 (m, 1H), 4.15-3.91 (m, 1H), 3.77-3.63 (m, 6H), 2.80-2.42 (m, 1H), 2.25-1.72 (m, 5H).

## 4.3.11. Methyl N-benzoyl-5-allyl-L-prolinate (7d)

Colorless oil;  $[\alpha]_D^{20}$  -26.9 (c=1.0 CHCl<sub>3</sub>); IR (neat)  $\nu$  = 2977, 2953, 1750, 1644, 1603, 1446, 1410, 1277, 1203, 1174, 1076 cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) (cis/trans = 13/87)  $\delta$  7.53-7.28 (m, 5H), 5.97-5.80 (m, 0.5H), 5.55-5.38 (m, 0.5H), 5.16-4.72 (m,

2H), 4.44-4.17 (m, 1H), 3.96 (br s, 0.5H), 3.77-3.60 (m, 3H), 3.06 (br s, 0.5H), 2.23-1.18 (m, 6H);  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>) (*cis/trans* = 13/87, a mixture of rotamers)  $\delta$  173.0, 171.5, 134.7, 133.5, 129.7, 128.3, 128.2, 126.7, 126.5, 118.1, 117.6, 62.1, 59.6, 59.3, 58.9, 52.2, 39.1, 38.4, 37.7, 28.9, 28.5, 26.8; HR-FAB(+) m/z calcd for  $C_{16}H_{20}NO_3[M+H]^+$  274.1443 found 274.1444.

### 4.4. Synthesis of cis-N-formyl-5-(2,4,6-triethylphenyl)-L-proline (4)

5% Pd-C (30 mg) was added to the solution of 6c (2.0 mmol, 847 mg) and triethylamine (279 µL, 2.0 mmol) in MeOH (5.0 mL). The mixture was then stirred under 1 atm of H<sub>2</sub> for 12 h. Upon completion of reaction the mixture was then filtered through celite and solvent removed obtain in vacuo to methyl cis-5-(2,4,6-triethylphenyl)-L-prolinate which was used for next reaction without further purification. Colorless oil;  $[\alpha]^{28}$ <sub>D</sub> +13.4 (c=1.1, CHCl<sub>3</sub>); IR (neat) v = 3350, 2963, 1734, 1458, 1210, 874, 669 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  6.89 (s, 2H), 4.57 (t, J= 8.7 Hz, 1H), 3.87 (t, J = 7.8 Hz, 1H), 3.76 (s, 3H), 2.90-2.50 (m, 6H), 2.35-2.00 (m, 4H), 1.78 (br s, 1H), 1.23 (t, J=7.5 Hz, 9H); HR-EI(+) m/z calcd for  $C_{18}H_{27}NO_2$  [M]<sup>+</sup> 289.2042, found 289.2027.

Under an argon atmosphere, acetic anhydride (2.0 mL) was added dropwise to a solution of methyl *cis*-5-(2,4,6-triethylphenyl)-L-prolinate in formic acid (6.0 mL) and stirred at room temperature for 9 h. Upon completion of reaction the solvent was removed under reduced pressure, then the residue was purified by silica gel column chromatography (*n*-hexane : AcOEt = 3 : 1) to afford methyl *cis-N*-formyl-5-(2,4,6-triethylphenyl)-L-prolinate<sup>4</sup> as a colorless crystal (372 mg, 58 % for 2 steps). Then, aqueous 1M NaOH (2.0 mL) was added to the stirred solution of

methyl *cis-N*-formyl-5-(2,4,6-triethylphenyl)-L-prolinate (1.0 mmol, 317 mg) in MeOH (4.0 mL), and the solution was stirred at room temperature for 12 h. The solution was neutralized with 3% aqueous HCl, and then MeOH was evaporated. The residue was diluted with brine, extracted with AcOEt, and dried over MgSO<sub>4</sub>. Removal of the solvent afforded compound  $\mathbf{4}^4$  (303 mg, quant.) as colorless crystals. Mp 132-133 °C;  $[\alpha]_D^{25}$ -135.5 (c=0.5, CHCl<sub>3</sub>); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.96 (s, 1H), 6.98 (s, 1H), 6.89 (s, 1H), 5.21 and 5.19 (d, J= 11.0 Hz, 1H), 4.75 (q, J= 9.3 Hz, 1H), 2.90 and 2.88 (d, J= 10 Hz, 1H), 2.85-2.05 (m, 9H), 1.30-1.10 (m, 9H).

4.5. Synthesis of N-picolinoyl (2S,5S)-[2,5-bis-(2,4,6-trimethylphenyl)]pyrrolidine (5)
4.5.1. trans-N-Benzoyl-5-(2,4,6-trimethylphenyl)-L-proline (8)

NaOH (12.9 mmol, 516 mg) was added to the stirred solution of **3d** (6.5 mmol, 2.27 g) in THF/H<sub>2</sub>O = 1 : 1 (60 mL), and the solution was stirred at room temperature for 4 h. The solution was then neutralized with 10% aqueous HCl, and extracted with AcOEt (150 mL x 3), and dried over MgSO<sub>4</sub>. After removal of the solvent and recrystallization from CHCl<sub>3</sub>/hexane, compound **8** was obtained as colorless crystals (1.27 g, 58%). Mp 204-207°C;  $[\alpha]_D^{19}$  -82.3 (c=0.3, CHCl<sub>3</sub>), IR (neat) v = 3640, 1727, 1642, 1620, 1445, 1354, 1123 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.20-7.00 (m, 5H), 6.66 (s, 1H), 6.41 (s, 1H), 5.38 (t, J= 8.4 Hz, 1H), 4.83 (t, J= 7.8 Hz, 1H), 4.20 (br s, 1H), 2.58-1.90 (m, 13H); <sup>13</sup>C NMR (100MHz, CDCl<sub>3</sub>) (a mixture of rotamers)  $\delta$  176.1, 171.4, 136.2, 136.0, 135.7, 135.1, 134.5, 134.1, 134.0, 131.2, 130.2, 129.2, 128.9, 128.1, 127.6, 127.4, 125.6, 63.1, 61.4, 60.3, 59.4, 32.3, 31.0, 29.9, 28.6, 20.6, 20.4, 20.3; EA calcd for C<sub>21</sub>H<sub>23</sub>NO<sub>3</sub>: C 74.75, H 6.87, N 4.15: found C 74.41, H 6.92, N 3.93; HR-EI(+) m/z calcd for C<sub>21</sub>H<sub>23</sub>NO<sub>3</sub> [M]<sup>+</sup> 337.1678, found 337.1674.

### 4.5.2. N-Benzoyl-2-methoxy-(5S)-(2,4,6-trimethylphenyl)pyrrolidine (9)

Anodic oxidation of **8** was carried out using graphite cathode (10 cm x 5 cm) and platinum anode (12 cm x 5 cm) in an undivided beaker-type cell. **8** (29.4 mmol, 9.9 g), and 2,6-lutidine (38.2 mmol, 4.5 mL) were added into MeOH (200 mL). After passing through 2.0 *F*/mol of electricity at constant voltage (18 V) at 0°C, MeOH was evaporated, then the residue was poured in water and extracted with AcOEt (200 mL x 3). The combined organic layer was dried over MgSO<sub>4</sub> and solvent removed under reduced pressure. The residue was purified by silica gel column chromatography (*n*-hexane : AcOEt = 3 : 1) to afford **9** (6.9 g, 73% yield) as colorless oil.  $[\alpha]_D^{24}$  +17.8 (*c*=1.0, CHCl<sub>3</sub>); IR (neat) v = 2732, 1765, 1727, 1692, 1642, 1613, 1582, 1547, 1503, 1468 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.83 (br s, 2H), 7.37 (br s, 3H), 6.77 (s, 2H), 5.23 (br s, 1H), 4.72 (br s, 1H), 3.14 (s, 3H), 2.60-2.03 (m, 13H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) (a mixture of diastereomers and rotamers)  $\delta$  171.3, 169.2, 135.8, 133.8, 133.4, 132.6, 132.0, 130.7, 129.7, 129.5, 127.8, 126.5, 125.9, 125.6, 125.2, 123.2, 92.7, 90.0, 60.2, 58.1, 54.5, 31.5, 31.1, 30.6, 28.6, 20.6, 20.5; HR-EI(+) *m/z* calcd for C<sub>21</sub>H<sub>25</sub>NO<sub>2</sub> [M]<sup>+</sup>323.1885, found 323.1866.

### 4.5.3. *N-Benzoyl-*(2S,5S)-[2,5-bis-(2,4,6-trimethylphenyl)]pyrrolidine (**10**)

Under an argon atmosphere,  $TiCl_4(140~\mu L, 1.0~mmol)$  was added dropwise to the solution of **9** (313 mg, 0.97 mmol) and 1,3,5-trimethylbenzene (400  $\mu$ L, 2.9 mmol) in  $CH_2Cl_2$  (10 mL) at -78°C. The resulting mixture was stirred for 24 h and allowed to stand until it warmed to room temperature. The solution was poured in ice water (10 mL) and extracted with  $CHCl_3$  (10 mL x 3). The combined organic layer was dried over

MgSO<sub>4</sub> and the solvent removed under reduced pressure. The residue was purified by silica gel column chromatography (n-hexane : AcOEt = 10 : 1) to afford **10** (351 mg, 88%) as colorless crystals. Mp 184-187 °C; [ $\alpha$ ]<sub>D</sub><sup>22</sup> +24.9 (c=0.5, CHCl<sub>3</sub>); IR (neat) v = 2963, 1738, 1632, 1580, 1483, 1408, 1348, 1240, 1102, 849, 795 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.20-7.00 (m, 5H), 6.85 (s, 1H), 6.81 (s, 1H), 6.60 (s, 1H), 6.35 (s, 1H), 5.65-5.59 (m, 2H), 2.61-2.20 (m, 22H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) (a mixture of rotamers)  $\delta$  169.7, 137.3, 136.8, 136.0, 135.8, 134.9, 134.6, 134.5, 133.6, 131.2, 131.0, 129.3, 129.2, 128.8, 127.1, 126.2, 60.3, 60.0, 32.4, 30.2, 21.2, 20.7, 20.6, 20.4, 20.1; EA calcd for C<sub>29</sub>H<sub>33</sub>NO: C 84.63, H 8.08, N 3.40: found C 84.43, H 8.15, N 3.02; HR-EI(+) m/z calcd for C<sub>29</sub>H<sub>33</sub>NO [M]<sup>+</sup> 411.2562, found 411.2560. HPLC: Daicel Chiralcel OD-H column, n-hexane : ethanol = 30 : 1, wavelength: 254 nm, flow rate: 1.0 mL/min, retention time: 8.9 min for (2R,5R)-10, 11.9 min for (2S,5S)-10.

## 4.5.4. N-Benzyl-(2S,5S)-[2,5-bis-(2,4,6-trimethylphenyl)]pyrrolidine (11)

1.03 M BH<sub>3</sub>-THF (17.4 mL, 18.0 mmol) was added to the solution of **10** (3.6 g, 8.7 mmol) in THF (70 mL), and refluxed at 80°C for 17 h. The solution was poured in water (100 mL) and extracted with AcOEt (100 mL x 3). The combined organic layer was dried over MgSO<sub>4</sub> and solvent removed in vacuo to obtain **11** (3.45 g, quant.), which was used for next reaction without further purification. Colorless crystal; mp  $109-110\,^{\circ}$ C; [ $\alpha$ ]<sub>D</sub><sup>23</sup> -124.5 (c=1.0, CHCl<sub>3</sub>); IR (neat) v = 2947, 1611, 1480, 1372, 1312, 1213, 1188, 1165, 1105, 1075, 1028, 851, 741, 700 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.00-6.89 (m, 3H), 6.78 (s, 2H), 6.63 (s, 2H), 6.38 (dd, J= 2.1, 7.8 Hz, 2H), 4.95 (t, J= 7.2 Hz, 2H), 3.37 (q, J= 12.9 Hz, 2H), 2.41-2.11 (m, 22H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  140.9, 138.8, 136.9, 136.3, 135.8, 131.2, 129.5, 129.2, 127.3, 125.9, 60.9,

51.7, 31.0, 21.5, 20.9; HR-EI(+) m/z calcd for  $C_{29}H_{35}N$  [M]<sup>+</sup> 397.2769, found 397.2766.

## 4.5.5. (2S,5S)-[2,5-Bis-(2,4,6-trimethylphenyl)]pyrrolidine (12)

20% Pd(OH)<sub>2</sub> (80 mg, 0.12 mmol) was added to the solution of **11** (228 mg, 0.57 mmol) and 3 drops of concentrated aqueous HCl in MeOH (5.0 mL). The mixture was then stirred under 1 atm of H<sub>2</sub> for 3 h. Upon completion of reaction the mixture was then filtered through celite and solvent removed in vacuo. The residue was poured into saturated aqueous NaHCO<sub>3</sub> (10 mL) and extracted with CHCl<sub>3</sub> (20 mL x 3). The combined organic layer was dried over MgSO<sub>4</sub> and solvent removed in vacuo to afford **12** (142 mg, 81% from **10**), which was used for next reaction without further purification. Colorless oil;  $[\alpha]_D^{22}$  -107.1 (c=0.5, CHCl<sub>3</sub>); IR (neat) v = 2951, 1611, 1462, 1084, 849 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  6.81 (s, 4H), 5.04 (t, J= 7.2 Hz, 2H), 2.46 (s, 12H), 2.23 (s, 6H), 2.13-2.08 (m, 4H), 1.68 (br s, 1H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  134.8, 134.2, 133.5, 128.2, 56.4, 31.0, 18.7, 18.6; HR-EI(+) m/z calcd for C<sub>22</sub>H<sub>29</sub>N [M]<sup>+</sup> 307.2300, found 307.2281.

### 4.5.6. *N-Picolinoyl-*(2*S*,5*S*)-[2,5-bis-(2,4,6-trimethylphenyl)]pyrrolidine (**5**)

A solution of picolinic acid (68.9 mg, 0.55 mmol) and CDI (122 mg, 0.75 mmol) in  $CH_2Cl_2$  (2.5 mL) was stirred at 0°C for 30 min. Then, a solution of **12** (153 mg, 0.50 mmol) in  $CH_2Cl_2$  (2.5 mL) was added at 0°C, and the mixture was stirred at room temperature for 24 h. The solution was poured into saturated aqueous NaHCO<sub>3</sub> (10 mL) and extracted with AcOEt (20 mL x 3). The combined organic layer was dried over MgSO<sub>4</sub> and solvent removed under reduced pressure. The residue was purified by silica gel column chromatography (n-hexane : AcOEt = 5 : 1) to afford **5** (192 mg, 93% yield)

as colorless crystals. Mp 73-74 °C;  $[\alpha]_D^{20}$  +6.8 (c=0.3, CHCl<sub>3</sub>), IR (neat) 2963, 1738, 1639, 1503, 1443, 1408, 1356, 1287, 1242, 1183, 1107, 851 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  8.30 (d, J= 4.2 Hz, 1H), 7.32-7.26 (m, 2H) 6.91 (t, J= 4.8 Hz, 1H), 6.85 (s, 1H), 6.79 (s, 1H), 6.53 (s, 1H), 6.40 (s, 1H), 6.03 (t, J= 7.2 Hz, 1H), 5.67 (t, J= 7.2 Hz, 1H), 2.63-2.02 (m, 22H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) (a mixture of rotamers)  $\delta$  167.1, 154.4, 146.7, 136.1, 135.9, 135.7, 135.5, 135.4, 134.6, 134.5, 133.9, 131.1, 130.7, 129.0, 128.6, 123.7, 122.5, 60.0, 59.8, 32.2, 29.9, 21.0, 20.6, 20.5, 20.3, 19.9; EA calcd for C<sub>28</sub>H<sub>32</sub>N<sub>2</sub>O: C 81.51, H 7.82, N 6.79: found C 81.21, H 7.84, N 6.54; HR-EI(+) m/z calcd for C<sub>28</sub>H<sub>32</sub>N<sub>2</sub>O [M]<sup>+</sup> 412.2515, found 412.2506.

#### 4.6. General procedure for asymmetric reduction of imines 13a-f

Cl<sub>3</sub>SiH (0.45 mmol) was added into a solution of imines **13a** (0.3 mmol) and compound **5** (0.03 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (1.5 mL), and the mixture was stirred at room temperature for 4 h. The mixture was then poured into saturated aqueous NaHCO<sub>3</sub> (10 mL) and extracted with CHCl<sub>3</sub> (10 mL x 3). The combined organic layer was dried over MgSO<sub>4</sub> and solvent removed under reduced pressure. The residue was purified by silica gel column chromatography to afford amine **14a** (159 mg, 77% yield).

## 4.6.1. (S)-N-Phenyl-N-(1-phenylethyl)amine (14a)<sup>6b</sup>

HPLC: Daicel Chiralcel OD-H column, n-hexane : isopropanol : diethylamine = 10 : 1 : 0.01, wavelength: 254 nm, flow rate: 1.0 mL/min, retention time: 7.2 min for (S)-14a, 8.6 min for (R)-14a.

## 4.6.2. (S)-N-[1-(4-Methoxylphenyl)ethyl-N-phenylamine ( $\mathbf{14b}$ ) $^{6b}$

HPLC: Daicel Chiralcel OD-H column, n-hexane : isopropanol = 99 : 1, wavelength: 254 nm, flow rate: 0.7 mL/min, retention time: 13.1 min for (S)-14b, 14.4 min for (R)-14b.

## 4.6.3. (14c) (S)-N-(4-Methoxylphenyl)-N-(1-phenylethyl)amine (14c)<sup>6b</sup>

HPLC: Daicel Chiralcel OD-H column, n-hexane : isopropanol = 99 : 1, wavelength: 254 nm, flow rate: 0.7 mL/min, retention time: 17.5 min for (S)-12c, 19.3 min for (R)-14c.

## 4.6.4. (S)-N-[1-(4-Chlorophenyl)ethyl]-N-phenylamine (14d)<sup>6c</sup>

HPLC: Daicel Chiralcel OD-H column, n-hexane: isopropanol = 95: 5, wavelength: 254 nm, flow rate: 1.0 mL/min, retention time: 9.0 min for (S)-12d, 10.8 min for (R)-14d.

## 4.6.5. (S)-N-[1-(4-Acetylphenyl)ethyl]-N-phenylamine (14e)

Pale yellow oil;  $[\alpha]_D^{27}$  -18.8 (c=0.7, CHCl<sub>3</sub>), IR (neat) 3390, 3054, 2980, 2926, 2869, 1678, 1603, 1506, 1429, 1360, 1320, 1269, 1210, 1181, 1144, 1015, 1015, 959 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.92 (d, J= 8.7 Hz, 2H), 7.47 (d, J= 8.7 Hz, 2H), 7.08 (t, J= 6.9 Hz, 2H), 6.65 (t, J= 6.3 Hz, 1H), 6.47 (d, J= 7.8 Hz, 2H), 4.53 (q, J= 8.2 Hz, 1H), 4.08 (br s, 1H), 2.58 (s, 3H), 1.53 (d, J= 6.9 Hz, 3H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  197.7, 151.0, 146.8, 136.0, 129.2, 129.1, 128.9, 126.0, 113.2, 53.4, 26.6, 24.9; HR-EI(+) m/z calcd for  $C_{16}H_{17}NO$  [M]<sup>+</sup> 239.1310, found 239.1287. HPLC: Daicel Chiralcel OD-H column, n-hexane: isopropanol = 5:1, wavelength: 254 nm, flow rate: 1.0 mL/min, retention time: 11.1 min for (S)-14e, 13.2 min for (R)-14e.

4.6.6. (S)-N-[1-(4-Nitrophenyl)ethyl]-N-phenylamine (14f)<sup>6b</sup>

HPLC: Daicel Chiralcel OD-H column, n-hexane: isopropanol = 95: 5, wavelength: 254 nm, flow rate: 1.0 mL/min, retention time: 33.5 min for (S)-14f, 38.0 min for (R)-14f.

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#### **References and notes**

- (a) Pichon, M.; Figadère, B. *Tetrahedron: Asymmetry* 1996, 7, 927-964. (b) Brenneman, J. B.; Machauer, R.; Martin, S. F. *Tetrahedron* 2004, 60, 7301-7314. (c) Esseveldt, B. C. J.; Vervoort, P. W. H.; van Delft, F. L.; Rutjes, P. J. T. *J. Org. Chem.* 2005, 70, 1791-1795. (d) Davis, F. A.; Song, M.; Augustine, A. *J. Org. Chem.* 2005, 71, 2779-2786.
- (a) Halland, N.; Braunton, A.; Bachmann, S.; Marigo, M.; Jørgensen, K. A. J. Am. Chem. Soc. 2004, 126, 4790-4791. (b) Simonini, V.; Benaglia, M.; Pignataro, L.; Guizzetti, S.; Celentano, G. Synlett 2008, 1061-1065.
- (a) Shono, T.; Matsumura, Y.; Tsubata, K. J. Am. Chem. Soc. 1981, 103, 1172-1176.
   (b) Shono, T.; Matsumura, Y.; Tsubata, K.; Uchida, K. J. Org. Chem. 1986, 51, 2590-2592.
   (c) Dhimane, H.; Vanucci-Bacqué, C.; Hamon, L.; Lhommet,

- G. Eur. J. Org. Chem. **1998**, 9, 1955-1963. (d) Kim, S.; Hayashi, K.; Kitano, Y.; Tada, M.; Chiba, K. Org. Lett. **2002**, 4, 3735-3737. e) Onomura, O.; Ishida, Y.; Maki, T.; Minato, D.; Demizu, Y.; Matsumura, Y. Electrochemistry **2006**, 74, 645-648.
- 4. Matsumura, Y.; Ogura, K.; Kouchi, Y.; Iwasaki, F.; Onomura, O. *Org. Lett.* **2006**, 8, 3789-3792.
- Iwasaki, F.; Onomura, O.; Mishima, K.; Maki, T.; Matsumura, Y. *Tetrahedron Lett.* 1999, 40, 7507-7511.
- (a) Iwasaki, F.; Onomura, O.; Mishima, K.; Kanematsu, T.; Maki, T.; Matsumura, Y. Tetrahedron Lett. 2001, 42, 2525-2527. (b) Malkov, A. V.; Mariani, A.; MacDougall, K. N.; Kočovský, P. Org. Lett. 2004, 6, 2253-2266. (c) Onomura, O.; Kouchi, Y.; Iwasaki, F.; Matsumura, Y. Tetrahedron Lett. 2006, 47, 3751-3754. (d) Malkov, A. V.; Stewart Liddon, A. J. P.; Ramírez-López, P.; Bendová, L.; Haigh, D.; Kočovský, P. Angew. Chem. Int. Ed. 2006, 45, 1432-1435. (e) Wang, Z.; Wei, S.; Wang, C.; Sun, J. Tetrahedron: Asymmetry 2007, 18, 705-709.
- (a) Shono, T.; Matsumura, Y.; Tsubata, K.; Sugihara, Y.; Yamane, S.; Kanazawa, T.;
   Aoki, T.; J. Am. Chem. Soc. 1982, 104, 6697-6703. (b) Shono, T.; Matsumura, Y.;
   Kanazawa, T.; Habuka, M.; Uchida, K.; Toyoda, K. J. Chem. Res. (S). 1984,
   320-321; J. Chem. Res. (M). 1984, 2873-2889. (c) Shono, T.; Fujita, T.; Matsumura,
   Y. Chem. Lett. 1991, 81-84. (d) Célimène, C.; Dhimane, H.; Lhommet, G.
   Tetrahedron 1998, 54, 10457-10468.
- 8. Stereoconfiguration of *trans-3d* was determined by the X-ray analysis. Crystallographic data for this structure have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication number CCDC 686483.

- Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB21EZ, UK; fax: +44(0) 1223 336033 or e-mail: deposit@ccdc.cam.ac.uk.
- 9. After hydrogenation of **7d**, its stereoconfiguration was determined by comparison with authentic sample, see: Cossy, J.; Cécile, D.; Pardo, D. G.; *Synlett* **1997**, 905-906.
- 10. Deprotection of **6b** with Me<sub>3</sub>SiI led to epimerization at the 5-position.
- (a) Iwasaki, T.; Horikawa, H.; Matsumoto, K.; Miyoshi, M. J. Org. Chem. 1979, 44,
   1552-1554. (b) Matsumura, Y.; Wanyoike, G. N.; Onomura, O.; Maki, T.
   Electrochim. Acta 2003, 48, 2957-2966.