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<tr>
<td>Citation</td>
<td>Chemical &amp; Pharmaceutical Bulletin v.49(4) p.486-487, 2001</td>
</tr>
<tr>
<td>Issue Date</td>
<td>2001-04</td>
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<tr>
<td>URL</td>
<td><a href="http://hdl.handle.net/10069/8399">http://hdl.handle.net/10069/8399</a></td>
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Revised Structure of Cercidinin A, a Novel Ellagitannin Having (R)-Hexahydroxydiphenoyl Esters at the 3,4-Positions of Glucopyranose

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Received October 30, 2000; accepted December 9, 2000

The structure of cercidinin A, an ellagitannin isolated from the bark of Cercidiphyllum japonicum, was revised to 1,2,6-tri-O-galloyl-3,4-(R)-hexahydroxydiphenoyl-β-D-glucose by two-dimensional NMR spectral analysis. Cercidinin A represents the first ellagitannin possessing a hexahydroxydiphenoyl group at the 3,4-positions of a modified C6-glucopyranose core.

Key words Cercidiphyllum japonicum; Cercidiphyllaceae; cercidinin A; tannin; ellagitannin

In the course of our chemical study on tannins, cercidinin A had been isolated from the bark of Cercidiphyllum japonicum Sieb. et Zucc. (Cercidiphyllaceae) and characterized as 1,4,6-tri-O-galloyl-2,3-(R)-hexahydroxydiphenoyl (HHDP)-β-D-glucose (1).3) Recently, Khanbabaee and Lötzereishere synthesized the 1,4,6-tri-O-galloyl-2,3-(R)-HHDP-β-D-glucose and pointed out that the NMR data of the product were not identical to those reported for cercidinin A.4) This paper and pointed out that the NMR data of the product were not identical to those reported for cercidinin A.4) This paper

Results and Discussion

In the previous work,3) cercidinin A was characterized by one-dimensional 1H- and 13C-NMR analysis and chemical examination. Methylation, followed by methanalysis, yielded (R)-dimethyl hexamethoxydiphenate ([α]D +27.1°) and methyl trimethoxybenzoate, unequivocally indicating the presence of a (R)-HHDP and galloyl groups in the molecule. Presence of a glucose core was verified by acid hydrolysis. The location of the esters on the glucopyranose was deduced from the result of selective hydrolysis of galloyl groups with tannase. Because of low resolution of the 1H-NMR spectra (100 MHz) in those days, complete assignment of the sugar proton signals of the hydrolysate, an anomer mixture of the tannase hydrolysate, 3,4-(R)-HHDP-glucose (1a), was also completely assigned, and a large low field shift of the signals, attributable to the glucose moiety. In the heteronuclear multiple bond correlation (HMBC) spectrum, signals attributable to glucose H-3 and H-4, confirmed the presence of a glucose core.

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In the present work, we first measured the 1H–1H correlation spectroscopy (1H–1H COSY) and the heteronuclear single quantum coherence (HSQC) spectrum of cercidinin A and made complete assignment for the proton and carbon signals arising from the glucose moiety. In the heteronuclear multiple bond correlation (HMBC) spectrum, signals attributable to glucose H-3 (δ 5.63, dd, J = 8.8, 10.1 Hz) and H-4 (δ 5.36, dd, J = 8.8, 10.1 Hz) were correlated with the ester carboxyl signals at δ 169.1 and 168.7, respectively. The ester carbons, in turn, coupled with the proton signals at δ 6.47 and δ 6.71 (each 1H, s), which were attributable to the aromatic protons of the HHDP group. Furthermore, the remaining carboxyl carbons of three galloyl groups, resonated at δ 165.0, 165.9, and 166.5, were coupled with the glucose H-1 (δ 6.28, d, J = 8.1 Hz), H-2 (δ 5.58, dd, J = 8.1, 10.1 Hz) and H-6 (δ 4.69, br d, J = 10.8 Hz, δ 4.54, dd, J = 4.2, 10.8 Hz), respectively. These HMBC correlations unambiguously clarified the location of the (R)-HHDP esters at the C-3 and C-4 positions of the glucose core. The 500 MHz 1H-NMR signals of the tannase hydrolysate, 3,4-(R)-HHDP-glucose (1a), were also completely assigned, and a large low field shift of the signals, attributable to the glucose moiety. In the heteronuclear multiple bond correlation (HMBC) spectrum, signals attributable to glucose H-3 and H-4, confirmed the presence of a glucose core. The 500 MHz 1H-NMR signals of the tannase hydrolysate, 3,4-(R)-HHDP-glucose (1a), were also completely assigned, and a large low field shift of the signals, attributable to the glucose moiety.

The 1H-NMR coupling constant between H-3 and H-4 (8.8 Hz) of 1 was slightly smaller than that observed for normal 4C1 glucopyranose (J3,4 = 9–10 Hz), suggesting that this glucose possessed a modified 4C1 conformation. In addition, the coupling patterns of the pyranose ring protons of 1a indicated that the pyranose conformation of the α-form (J3,4 = 10.3 Hz, J3,5 = 9.4 Hz, J4,5 = 10.5 Hz) was slightly different from that of the β-form, taking 4C1 conformation (J3,4 = J3,5 = J4,5 = 9.7 Hz).
Most monomeric ellagitannins are structurally and biogenetically classified into two groups.\textsuperscript{5} Tannins, belonging to the first group (group A), have the $\mathrm{C}_4$ glucopyranose core, and usually the (S)-HHDP esters are located at the 2,3-positions and/or 4,6-positions of the pyranose ring. Cercidinin A belongs to the diphenoyl esters (or its oxidized form, dehydrohexahydroxydiphenoyl esters) are attached to the 1,6-; 1,3-; 2,4-; or 3,6- positions of the pyranose ring.\textsuperscript{7} HHDP group.

**Experimental**

**General** $^1$H- and $^13$C-NMR spectra were obtained with Varian Unity plus 500 spectrometer operating at 500 MHz for $^1$H, and 125 MHz for $^13$C, respectively. the HMBC experiment (J\textsubscript{CH} optimized for 8 Hz) was performed using standard Varian pulse sequences.

**Cercidinin A (1)** An off-white amorphous powder, $[\alpha]_D^20 = -71.6^\circ$ ($c = 1.0$, acetone), FAB-MS $m/z$: 961 (M+Na)$^+$. $^1$H-NMR (500 Hz, acetone-$d_6$, $\delta$), 1H-NMR (500 Hz, acetone-$d_6$, $\delta$) and $^13$C-NMR (125 MHz, acetone-$d_6$) $\delta$: 169.1 [COO (connected to glucose C-3)], 168.7 [COO (C-4)], 166.5 [COO (C-6)], 165.9 [COO (C-2)], 165.0 [COO (C-1)], 146.1, 146.0 [galloyl-3, 5 (C-1, 2)], 145.9 [galloyl-3, 5 (C-6)], 145.2, 145.1 (HHDP-4, 4'), 144.4, 144.3 (HHDP-6, 6'), 139.8, 139.5 [galloyl-4 (C-1, 2)], 139.0 [galloyl-4 (C-6)], 136.4, 136.4 (HHDP-5, 5'), 126.1, 126.0 (HHDP-2, 2'), 120.9 [galloyl-1 (C-6)], 120.16, 119.5 [galloyl-1 (C-1, 2)], 114.6, 114.5 (HHDP-1, 1'), 110.2, 110.0 (galloyl-2, 6 (C-1, 2)], 109.9 [galloyl-2, 6 (C-6)], 107.5 (HHDP-3'), 107.3 (HHDP-3), 93.5 (C-1), 77.0 (C-3), 72.7 (C-5), 72.6 (C-4), 70.5 (C-2), 62.5 (C-6).

**Tannase Hydrolysis** Compound 1 (45 mg) in water (3 ml) was stirred with tannase at room temperature for 4 h. The mixture was directly subjected to Sephadex LH-20 column chromatography (1.5 cm i.d. x 25 cm) with water containing increasing proportions (0% to 40%) of MeOH to afford gallic acid (23.5 mg) and 1a (20.0 mg) as a white amorphous powder, $[\alpha]_D^20 = +12.9^\circ$ ($c = 1.0$, MeOH), $^1$H-NMR (500 Hz, acetone-$d_6$, $\delta$) and $^13$C-NMR (125 MHz, acetone-$d_6$) $\delta$: 6.72, 6.61 (each s, HHDP-H), 5.34 (dd, $J = 9.4$, 10.3 Hz, H-3), 5.31 (d, $J = 3.7$ Hz, H-1), 4.91 (dd, $J = 9.4$, 10.5 Hz, H-4), 4.16 (ddd, $J = 2.3$, 5.5, 10.5 Hz, H-5), 3.80 (dd, $J = 3.7$, 10.3 Hz, H-2), 3.77 (ddd, $J = 2.3$, 12.1 Hz, H-6a), 3.68 (dd, $J = 5.5$, 12.1 Hz, H-6b); $\beta$ form $\delta$: 6.70, 6.61 (each s, HHDP-H), 5.05 (t, $J = 9.7$ Hz, H-3), 4.91 (t, $J = 9.7$ Hz, H-4), 4.75 (d, $J = 7.6$ Hz, H-1), 3.81 (ddd, $J = 2.3$, 12.1 Hz, H-6a), 3.76 (ddd, $J = 2.3$, 5.5, 9.7 Hz, H-5), 3.66 (ddd, $J = 5.5$, 12.1 Hz, H-6b), 3.58 (dd, $J = 7.6$, 9.7 Hz, H-2); ($\alpha$, $\beta$ molar ratio = 3:2).

**References and Notes**

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